### Part B

# Systematic investigation in chemistry

#### 11. Execution (Time 30 minutes)

You are going to carry out an investigation to find out: Which one of the three solids common salt, bicarbonate or sugar has the best neutralizing capacity when they are mixed in an acidic solution?

#### In your execution, you are going to:

- work according to your plan or the prepared experiment instruction.
- consider the safety instructions your teacher has informed you about.
- take notes on your results in a table.



# Part B

### Experiment instruction for a systemic investigation in chemistry

You are going to carry out an investigation to find out:

Which one of the three solids common salt, bicarbonate or sugar has the best neutralizing capacity when they are mixed in an acidic solution?

#### **Equipment:**

Common salt, bicarbonate, sugar, a hydro chloric solution (prepared by your teacher), BTB, plastic cups or beakers, paper bun cases, spoons, graduated glass cylinder, goggles and apron.

#### **Risks with the experiment:**

Consider the safety instructions informed by your teacher.

#### Method of investigation:

- 1. Mark 3 beakers with **A**, **B** and **C**.
- 2. Measure 20 ml of the hydro chloric solution and pour in beaker **A**, beaker **B** and beaker **C**.
- **3.** Add BTB in beaker **A**, beaker **B** and beaker **C**.
- 4. Note and write down your observations.
- 5. Measure common salt, bicarbonate and sugar.
- 6. Add common salt in beaker A.
- 7. Add bicarbonate in beaker **B**.
- 8. Add sugar in beaker C.
- 9. Take notes on your results in a table.

# Part B

# Systematic investigation in chemistry

### 12. Evaluation (Time 30 minutes)

a) Present your results for the three compounds in a table.

- b)
- Draw a conclusion, from your results, on which of the three compounds that has the best neutralizing capacity.
- Explain the conclusion using your chemistry content knowledge.

Suppose you can do the investigation once again.

- c)
- Give **one** suggestion on an improvement of your investigation to make your result more reliable.
- Explain why this improvement would give a more reliable result.

